Table V-Dopa Response Potentiation

$$R_{3}$$
 — $CH_{2}CH_{2}$ — $CONR_{1}R_{2}$ — $CONR_{1}R_{2}$ — I_{1} — CH_{3}

Number	R ₃	$\mathbf{NR}_1\mathbf{R}_2$	Oral, mg./kg.	Motor Activity
IIIa	3,4-(OCH ₃) ₂	$N(C_2H_5)_2$	25	1+
IIIb	3,4-(OCH ₃) ₂	$N(C_2H_5)_2$	100 25	$\frac{1+}{3+}$
	, ,		100	3+
IIIc	3,4,5-(OCH ₃) ₃	$N(C_2H_5)_2$	25	2+
IIId	н	C ₄ H ₈ NO ^a	100 25	2+ 2+
IIIe	3,4-(OCH ₃) ₂	C ₄ H ₈ NO ^a	100 25	3+ 3+
	0,1 (0 0113)2	041-0110	100	2+
Шf	3,4,5-(OCH ₃) ₃	C ₄ H ₈ NO ^a	25	1+
Amitrip	tyline	100 20	$\frac{2+}{3+}$	

^a Morpholino.

 $(3 \times 50 \text{ ml.})$. After drying over anhydrous magnesium sulfate and evaporation of the solvent, the product was purified by elution chromatography, using neutral alumina I and anhydrous ether as the eluent.

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▲ To whom inquiries should be directed.

Chemistry and Biological Activity of N^1 -Acyl-4-arylazopyrazoles

H. G. GARG*▲ and VEENA ARORA

Keyphrases \square N¹-Acyl-4-arylazopyrazoles—synthesis, screened for pharmacological activity \square Pyrazole derivatives—synthesis of N¹-acyl-4-arylazopyrazoles, screened for pharmacological activity

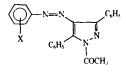
In continuation of work on the synthesis of pyrazoles as potential biologically active agents, preparation of N^1 -acyl analogs was undertaken since heterocyclics having a -- CONH₂ group possess various activities (1).

 N^{1} -Acetyl-3,5-diphenyl-4-arylazo-, N^{1} -acetyl-3,5-dimethyl-4-arylazo-, and N^{1} -sulfamoylbenzoyl-3,5dimethyl-4-arylazopyrazoles were prepared by the cyclization of 2-arylhydrazonopropane-1,3-diphenyl-1,2,3-triones (2) and 3-arylhydrazono-2,3,4-pentanetriones (3) with acetylhydrazine (4) or 4-sulfamoylbenzoylhydrazine (5). The yields ranged from 45 to 65%.

The IR (KBr) spectrum revealed bands characteristic of the acetyl group, the -N=N- grouping, and the C=C-NH-N= grouping in the regions 1684-1709, 1400-1540, and 1556-1565 cm.⁻¹, respectively, which are in good agreement with the assigned structures.

UV spectra of N1-sulfamoyl-3,5-dimethyl-4-(2,4-

Abstract [] The preparation of N¹-acyl-4-arylazopyrazole derivatives by cyclization of 3-arylhydrazono-2,3,4-pentanetriones or 2arylhydrazonopropane-1,3-diphenyl-1,2,3-triones with acetylhydrazine or 4-sulfamoylbenzoylhydrazine is presented. The structure was assigned on the basis of elemental analysis and IR data. No significant activity was observed against Gram-positive and Gramnegative bacteria, fungi, and *Trichomonas foetus* P-1005.



Number	x	Yield, %	Melting Point	Colorª	Formula	Calc.	is, % Found
1	Н-—	60	155°	OSp	C ₂₃ H ₁₈ N ₄ O	C 75.41 H 4.92 N 15.30	75.38 4.95 15.26
2	2-Br—	55	230°	PeYSp	$C_{23}H_{17}BrN_4O$	N 17.98 Br 12.58	18.00
3	2,5-(Me) ₂	65	195°	ON	$C_{25}H_{22}N_4O$	N 14.21	14.25
4	2,6-(Me)2	50	171°	GP	C25H22N4O	N 14.21	14.26
5	2-Et	50 55	193°	YN	C25H22N4O	N 14.21	14.25
6	2.4-(Me) ₂	55	150°	OYN	$C_{25}H_{22}N_4O$	N 14.21	14.23
7	2.3-(Me)2	60	232°	GYN	$C_{25}H_{22}N_4O$	N 14.21	14.25
8	2,5-(MeO) ₂	65	1 7 4°	BP	$C_{25}H_{22}N_4O_3$	N 13.14	13.18
9	2,5-(EtO)2	55	172°	ON	$C_{27}H_{26}N_4O_3$	N 12.33	12.35
10	2,4-(Cl) ₂	50	222°	GN	$C_{23}H_{16}Cl_2N_4O$	N 12.87 Cl 16.32	12.90 16.28
11	4-C1; 2,5- (MeO)2	55	184°	DYN	$C_{25}H_{21}ClN_4O_3$	N 12.16 Cl 7.71	12.12 7.74

Table I--Characteristics of N1-Acetyl-3,5-diphenyl-4-arylazopyrazoles

^a O = orange, Sp = specks, Pe = pale, Y = yellow, N = needles, G = golden, B = brown, D = dark, M = mustard, and Br = bright.

Table IICharacteristics of N ¹ -Acetyl-3,5-dimethyl-4-arylazopyrazoles Yield, Melting							COCH ₃	
Number	Х	%	Point	Color ^a	Formula	Calc.	s, % Found	
1	2-Me	55	196°	GP	C14H16N4O	C 65.63 H 6.30 N 21.88	65.60 6.34 21.90	
2 3 4	3-MeO 3,4-(Me) ₂ 2,3-(Cl) ₂	60 55 40	170° 195° 201°	MSp BrYN ON	C ₁₄ H ₁₆ N ₄ O ₂ C ₁₅ H ₁₈ N ₄ O C ₁₃ H ₁₂ Cl ₂ N ₄ O	N 20.59 N 20.74 N 18.01 Cl 22.83	20.63 20.70 17.96 22.85	

^a See footnote of Table I.

dichlorophenylazo)pyrazoles and N^1 -sulfamoylbenzoyl-3,5-dimethyl-4-(2-methoxyphenylazo)pyrazoles showed maxima at 242/360 and at 244/330 nm., respectively.

PHARMACOLOGY

The compounds prepared in this study were tested for antibacterial activity *in vitro*, using the agar diffusion method. Five mice were given a single oral dose equivalent to 500 mg./kg. Pooled 5-hr. urine was then tested for activity using the agar diffusion method. Filter paper disks (6.3 mm.) were dipped in a 20-mg./ml. suspension of the test compound and tested for activity.

EXPERIMENTAL¹

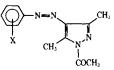
3-Arylhydrazono-2,3,4-pentanetriones (2) and 2-arylhydrazonopropane-1,3-diphenyl-1,2,3-triones (3) were prepared by earlier described procedures. Acetylhydrazine—This hydrazine was synthesized by treating acetyl chloride with hydrazine hydrate in equimolar quantities and keeping the mixture overnight at 0° . The resultant solid was recrystallized from ethanol (m.p. 67°).

4-Sulfamoylbenzoylhydrazine -p-Toluenesulfonamide was oxidized according to the method of Kamm and Mathews (6). The acid obtained was esterified, and the ester formed was treated with hydrazine hydrate to yield the solid hydrazide, which was finally recrystallized from boiling water (m.p. 238°).

 N^1 -Acetyl-3,5-diphenyl-4-arylazopyrazoles – A solution of acetylhydrazine (0.005 mole) in alcohol (15 ml.) containing a few drops of concentrated H₂SO₄ was added to the appropriate 2-arylhydrazonopropane-1,3-diphenyl-1,2,3-trione (0.005 mole) dissolved in an alcohol-acetic acid mixture. The resultant solution was boiled under reflux for several hours and then cooled. The reaction mixture was diluted with water, and the crystals which separated were collected and recrystallized from ethanol or acetic acid. The characteristics of N^1 -acetyl-3,5-diphenyl-4-arylazopyrazoles are listed in Table I.

 N^1 -Acetyl-3,5-dimethyl-4-arylazopyrazoles – These were obtained from acetylhydrazine (0.005 mole) and 3-arylhydrazono-2,3,4pentanetrione (0.005 mole) by the same procedure as was used for N^1 -acetyl-3,5-diphenyl-4-arylazopyrazoles. The yields and physical constants of these pyrazoles are listed in Table II.

 N^{1} -4-Sulfamoylbenzoyl-3,5-dimethyl- 4 - arylazopyrazoles—Treatment of 3-arylhydrazono-2,3,4-pentanetrione (0.005 mole) with 4sulfamoylbenzoylhydrazine under conditions similar to those used



¹ Analyses were done at C.D.R.I., Lucknow, India. Melting points were determined on a Koffer hot-stage apparatus and are uncorrected. IR spectra were recorded on a Perkin-Elmer Infracord in the KBr phase. UV spectra were measured for solutions in EtOH.

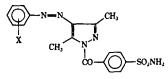


Table III—Characteristics of N ¹ -(4'-Sulfamoylbenzoyl)-3,5-dimethyl-4-aryla

Number			Melting		Formula	Analysis, %	
	х		Point			Calc.	Found
1	Н—	55	245° dec.	BN	$C_{18}H_{17}N_{5}O_{3}S$	C 56.40 H 4.44 N 18.28 S 5.35	56.38 4.48 18.30 5.34
2	2-Me	55	188°	MSp	$C_{19}H_{19}N_3O_3S$	N 17.63 S 8.06	17.65 8.08
3	3-MeO	50	160°	GP	$C_{19}H_{19}N_5O_4S$	N 16.94 S 7.75	16.95 7.78
4	4-MeO	60	25 0°	DBP	$C_{19}H_{19}N_5O_4S$	N 16.94 S 7.75	16.90 7.76
5	3,4-(Me)2	55	24 0°	YSp	$C_{20}H_{21}N_5O_3S$	N 17.03 S 7.79	17.05 7.78
6	2,4-(Cl) ₂	45	250°	DYSp	$C_{18}H_{15}Cl_2N_5O_3S$	N 15.45 S 7.06	15.48 7.08
7	2,3-(Cl) ₂	50	250°	MN	$C_{18}H_{15}Cl_2N_5O_8S$	N 15.45 S 7.06	15.44 7.08
8	2,6-(Cl) ₂	55	189°	BP	$C_{18}H_{15}Cl_2N_5O_3S$	N 15.45 S 7.06	15.46 7.04
9	5-Cl; 2,4- (MeO) ₂	55	235°	MYSp	$C_{20}H_{20}CIN_6O_5S$	N 13.65 S 6.24	13.68 6.22

^a See footnote of Table I.

in other cases gave pyrazoles, whose characteristics are listed in Table III.

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* Present address: Harvard University Medical School, Laboratory for Carbohydrate Research, Massachusetts General Hospital, Boston, MA 02114

▲ To whom inquiries should be directed.